

Highly Diastereoselective Enolate Addition of O-Protected α -Hydroxyacetate to (S_R)-tert-Butanesulfinylimines: Synthesis of Taxol Side Chain

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The taxol side chain ($S_R, 2R, 3S$)-*N*-tert-butanesulfinyl-*O*-Boc-3-phenylisoserine benzyl ester **4c** was synthesized through a lithium enolate addition of *O*-Boc- α -hydroxyacetate benzyl ester **5c** to benzylidene (S_R)-tert-butanesulfinamide **6a** in excellent yield and diastereoselectivity. By similar approach, a series of enantiopure 3-substituted isoserine benzyl esters **4** useful for the semi-syntheses of taxol derivatives were also prepared in high to excellent yields and diastereoselectivities. The diastereoselective addition mechanism was discussed on the basis of the experimental observation.

Introduction

Taxol 1,¹ isolated from *Taxus brevifolia*, is considered the most promising anticancer drug. Considering the complexity of taxol and the fact that the content of taxol in the bark of *T*. *brevifolia* is only about 0.03%, the major barrier for medical usage of taxol is resource limitation. Mass production by total synthesis² and by isolation from the natural resource no doubt is not economical and has sacrificed the rare species of *T*.

brevifolia. For this reason, extensive efforts have been focused on semi-synthesis³ of taxol by the condensation of commercially available 14-β-hydroxy-10-deacetylbacctin III **2** with a side chain such as *N*-benzoyl-(2*R*, 3*S*)-3-phenylisoserine **3** (Figure 1). Therefore, over the last 20 years, the efficient synthesis of enantiopure side chain **3** has attracted much attention from academic community as well as industry.⁴ In this article, we report a highly diastereoselective enolate addition of *O*-Bocα-hydroxyacetate to benzylidene (*S_R*)-*tert*-butanesulfinamide to lead directly to the formation of *N*,*O*-protected (*S_R*,*2R*,3*S*)-3phenylisoserine ester **4c** in excellent yield. The efficiency and applicable scope of *tert*-butanesulfinylimine enolate addition to prepare a series of enantiopure 3-substituted isoserines **4** were well demonstrated.

^{(1) (}a) Horwitz, S. B. J. Nat. Prod. **2004**, 67, 136. (b) Miller, M. L.; Ojima, I. Chem. Rec. **2001**, 1, 195. (c) Kingston, D. G. I. Chem. Commun. **2001**, 867. (d) Wani, M. C.; Taylor, H. I.; Wall, M. E.; Coggon, P.; McPhail, A. T. J. Am. Chem. Soc. **1971**, 93, 2325.

^{(2) (}a) Kusama, H.; Hara, R.; Kawahara, S.; Nishimori, T.; Kashima, H.; Nakamura, N.; Morihira, K.; Kuwajima, I. J. Am. Chem. Soc. 2000, 122, 3811.
(b) Mukaiyama, T.; Shiina, I.; Iwadare, H.; Saitoh, M.; Nishimura, T.; Ohkawa, N.; Sakoh, H.; Nishimura, K.; Tani, Y.; Hasegawa, M.; Yamada, K.; Saitoh, K. Chem.-Eur. J. 1999, 5, 121.
(c) Wender, P. A.; Badham, N. F.; Conway, S. P.; Floreancig, P. E.; Glass, T. E.; Granicher, C.; Houze, J. B.; Janichen, J.; Lee, D.; Marquess, D. G.; McGrane, P. L.; Meng, W.; Mucciaro, T. P.; Muhlebach, M.; Natchus, M. G.; Paulsen, H.; Rawlins, D. B.; Satkofsky, J.; Shuker, A. J.; Sutton, J. C.; Tayler, R. E.; Tomooka, K. J. Am. Chem. Soc. 1997, 119, 2755.
(d) Master, J. J.; Link, J. T.; Snyder, L. B.; Young, W. B.; Danishefsky, S. J. Angew. Chem., Int. Ed. 1995, 34, 1723.
(e) Holton, R. A.; Somoza, C.; Kim, H.-B.; Liang, F.; Biediger, R. J.; Boatman, P. D.; Shindo, M.; Smith, C. C.; Kim, S.; Nadizadeh, H.; Suzuki, Y.; Tao, C.; Vu, P.; Tang, S.; Zhang, P.; Murthi, K. K.; Gentile, L. N.; Liu, J. H. J. Am. Chem. Soc. 1994, 116, 1597.
(f) Nicolaou, K. C.; Yang, Z.; Liu, J. J.; Ueno, H.; Nantermet, P. G.; Guy, R. K.; Claiborne, C. F.; Renaud, J.; Couladouros, E. A.; Paulvannan, K.; Sorensen, E. J. Nature 1994, 367, 630.

^{(3) (}a) Gennari, C.; Vulpetti, A.; Donghi, M.; Mongelli, N.; Vanotti, E. Angew. Chem., Int. Ed. **1996**, *35*, 1723. (b) Ojima, I. Acc. Chem. Res. **1995**, 28, 383. (c) Ojima, I.; Habus, I.; Zhao, M.; Zucco, M.; Park, Y. H. Tetrahedron **1992**, *48*, 6985.

^{(4) (}a) Tosaki, S.; Tsuji, R.; Ohshima, T.; Shibasaki, M. J. Am. Chem. Soc. 2005, 127, 2147. (b) Kudyba, I.; Raczko, J.; Jurczak, J. J. Org. Chem. 2004, 69, 2844. (c) Juhl, K.; Jorgensen, K. A. J. Am. Chem. Soc. 2003, 124, 2420. (d) Aggarwal, V. K.; Vasse, J. L. Org. Lett. 2003, 5, 3987. (e) Schade, W.; Reissig, H. U. J. Prakt. Chem. 1999, 341, 685. (f) Li, G.; Chang, H.-T.; Sharpless, K. B. Angew. Chem., Int. Ed. 1996, 35, 451. (g) Ojima, I. Acc. Chem. Res. 1995, 28, 383. (h) Wang, Zh.-M.; Kolb, H. C.; Sharpless, K. B. J. Org. Chem. 1994, 59, 5104. (i) Crispino, G. A.; Jeong, K. S.; Kolb, H. C.; Wang, Zh.; Xu, D. J. Org. Chem. 1993, 58, 3785. (j) Hattori, K.; Miyata, M.; Yamamoto, H. J. Am. Chem. Soc. 1993, 115, 1151. (k) Deng, L.; Jacobsen, E. N. J. Org. Chem. 1992, 57, 4320. (l) Denis, J. N.; Correa, A.; Greene, A. E. J. Org. Chem. 1990, 55, 1957.



FIGURE 1. Structure of taxol, bacctin III, and taxol side chains.







In recent years, Davis, Ellman, and co-workers have demonstrated well the diastereoselective sulfinylimine enolate addition reaction as one of the important reactions of C–C bond formation for preparing chiral β -amino acid ester and α , β diamino acid ester.^{5,6} In continuing our study on chiral *tert*butanesulfinamide chemistry,⁷ we envisioned that enantiopure 3-phenylisoserine ester **4** with a (2*R*,3*S*) configuration may be directly assembled by the diastereoselective addition of an enolate of *O*-protected α -hydroxyacetate **5** to enantiopure *tert*butanesulfinylimine **6a**, although theoretically four isomers will be formed in the addition reaction (Scheme 1).

Initial experiments showed that the addition reaction of imine **6a** with a series of lithium enolates of benzyl acetates **5** ($R^1 = Et_3Si$, 'BuMe₂Si, Bn, *p*-MeOC₆H₄, 'BuCO, and Bz, $R^2 = Bn$), generated by the reaction of **5** with LDA at -78 °C in THF, respectively, gave unexceptionally a complicated mixture in low yield. When the hydroxyl-protecting group of benzyl acetate **5** was changed to a carbonate group (**5a**, $R^1 = MeOCO$, $R^2 = Bn$), we were pleased to find that the major *syn*-adduct **4a** with a (*2R*,*3S*) configuration was isolated in 45% yield and moderate diastereoselectivity (Table 1, entry 1).⁸ With the size increase of R^1 from methoxy carbonyl group to *tert*-butoxy carbonyl group, the yield and diastereoselectivity were greatly improved (Table 1, entries 1–3). Only one adduct, **4c**, was observed and

(7) (a) Huang, Zh. Y.; Zhang, M.; Wang, Y.; Qin, Y. Synlett 2005, 1334.
(b) Ke, B.; Qin, Y.; He, Q. F.; Huang, Zh. Y.; Wang, F. P. Tetrahedron Lett. 2005, 46, 1751. (c) Qin, Y.; Wang, C.; Huang, Zh. Y.; Xiao, X.; Jiang, Y. Zh. J. Org. Chem. 2004, 69, 8533.

(8) The syn-adducts with a typical double doublet ascribed to H-3 were readily differentiated from the anti-adducts with a typical triplet in the 1 H NMR spectrum.

TABLE 1. Diastereoselective Addition of 5 to (S_R) -N-(Benzylidene)-tert-butanesulfinamide 6a^a

entry	acetate 5	\mathbb{R}^1	\mathbb{R}^2	base	yield of 4^{b}	dr
1	5a	MeOCO	Bn	LDA	4a (45)	88:12:0:0 ^c
2	5b	Cbz	Bn	LDA	4b (56)	d
3	5c	Boc	Bn	LDA	4c (86)	>99:0:0:0 ^e
4	5c	Boc	Bn	BuLi	4c (36)	>99:0:0:0 ^e
5	5c	Boc	Bn	'BuLi	4c (43)	>99:0:0:0 ^e
6	5c	Boc	Bn	LiHMDS	4c (97)	>99:0:0:0 ^e
7 ^f	5c	Boc	Bn	LiHMDS	4c (50)	>99:0:0:0 ^e
8^g	5c	Boc	Bn	LiHMDS	4c (85)	>99:0:0:0 ^e
9	5d	Boc	Me	LiHMDS	4d (96)	>99:0:0:0 ^e

^{*a*} All reactions were conducted at -78 °C in THF using 1 equiv of **6a**, 5 equiv of **5**, and 5 equiv of base unless specifically indicated. ^{*b*} Isolated yield. In parentheses, percent value. ^{*c*} The dr value was based on isolated isomers, and the absolute configuration of minor isomer was not identified. ^{*d*} Other isomers were isolated as an inseparable and unidentified mixture contaminated with byproducts. ^{*e*} A dr value of >99% indicated that other isomers were not observed. ^{*f*} At -45 °C. ^{*g*} At -78 °C, 3 equiv of **5c** and LiHMDS were used.





 a Reagents and conditions: (a) 6 N HCl in MeOH, 1 h, 96%. (b) PhCOCl/ NaHCO_3, aq. THF, 85%.

was isolated in 86% yield when acetate **5c** ($\mathbb{R}^1 = \operatorname{Boc}$, $\mathbb{R}^2 = \operatorname{Bn}$) was used. After screening a couple of bases (Table 1, entries 3–6), the best result was achieved using 5 equiv of LiHMDS as a base and 5 equiv of **5c** at -78 °C in THF. Under the best condition, we were able to isolate **4c** ($\mathbb{R}^1 = \operatorname{Boc}$, $\mathbb{R}^2 = \operatorname{Bn}$) in 97% yield as sole adduct (Table 1, entry 6). Attempts to raise the reaction temperature and reduce the quantity of lithium enolate of **5c** unfortunately resulted in lower yields, although the diastereoselectivities were kept the same (Table 1, entries 7 and 8). Methyl acetate **5d** gave a similar result compared with benzyl acetate **5c** (Table 1, entry 9).

The absolute stereochemistry of major adduct **4** was verified by converting **4d** into the known methyl ester of (2*R*,3*S*)-3benzoyl-3-phenylisoserine (–)-**8** with the same rotation and identical ¹H NMR spectrum compared with that of the literature data (Scheme 2). (–)-**8**, mp: 183–184 °C, $[\alpha]^{25}_{D}$ –48.8 (*c* 1.25, MeOH) {lit.^{1d} $[\alpha]^{23}_{D}$ –49.6 (MeOH), lit.⁹ $[\alpha]^{23}_{D}$ –48.0 (MeOH)}. The absolute configuration of the major adduct was further unambiguously confirmed by the X-ray crystallographic analysis of **4c** as *S*_R,2*R*,3*S*.¹⁰

It was interesting to find the applicable scope and efficiency of current imine enolate addition strategy for the preparation of enantiopure isoserine **4** with different substituents at position-3, since taxol analogues with a variety of substituents instead of phenyl group at position-3 of the side chain have shown equal or even better anticancer profile, and some of them are being studied in different stages of clinical trials.¹¹

⁽⁵⁾ For reviews on chiral sulfinylimine chemistry, see: (a) Zhou, P.; Chen, B. C.; Davis, F. A. *Tetrahedron* **2004**, *60*, 8003. (b) Ellman, J. A.; Owens, T. D. *Pure Appl. Chem.* **2003**, *75*, 39. (c) Ellman, J. A.; Owens, T. D.; Tang, T. P. *Acc. Chem. Res.* **2002**, *35*, 984. (d) Davis, F. A.; Zhou, P.; Chen, B. C. *Chem. Soc. Rev.* **1998**, *27*, 13.

^{(6) (}a) Davis, F. A.; Deng, J. Org. Lett. **2004**, *6*, 2789. (b) Davis, F. A.; Prasad, K. R.; Nolt, M. B.; Wu, Y. Org. Lett. **2003**, *5*, 925. (c) Evans, J. W.; Ellman, J. A. J. Org. Chem. **2003**, *68*, 9948. (d) Tang, T. P.; Ellman, J. A. J. Org. Chem. **2002**, *67*, 7819.

⁽⁹⁾ Denis, J. N.; Greene, A. E.; Serra, A. A.; Luche, M. J. J. Org. Chem. **1986**, *51*, 46.

⁽¹⁰⁾ A colorless crystal of **4c** ($C_{25}H_{31}N_1O_6S_1$, mp 148–149 °C) suitable for X-ray analysis was obtained by crystallization from acetone/petroleum 1:3. The crystallographic data have been deposited with the Cambridge Crystallographic Data Centre, CCDC 276473, and are also available in Supporting Information (the ORTEP/X-ray figure of **4c** is shown in Supporting Information).

TABLE 2. Diastereoselective Addition of 5c to (S_R) -tert-Butanesulfinylimines 6



^{*a*} Isolated yield. ^{*b*} Inseparable mixture of two isomers in 11:1 ratio. ^{*c*} The ratio was based on ¹H NMR integration of crude product; a dr value of >99% indicated that other isomers were not observed, and the absolute configuration of minor isomer was not identified. ^{*d*} Other isomers were isolated as an inseparable and unidentified mixture contaminated with byproducts.

As shown in Table 2, the additions of 5c to both aromatic and alkyl imines 6 were completed within 6 h to afford the desired syn-4 as the major adduct in high to excellent yields and diastereoselectivities under our best conditions. The ratio was determined by careful ¹H NMR analyses of the crude product and each purified minor component. For most of the addition reactions, the reactions were very clean and only one adduct was detected by TLC analysis after quenching the reactions. The reactions with electron-rich imines such as **6c**, 6d, 6j, and 6k proceeded rapidly and were completed within 1 h (Table 2, entries 2, 3, 9, and 10), whereas the electron-poor imines 6f-6i exhibited a lower reactivity and required a longer reaction time (4-6 h) (Table 2, entries 5-8). For the addition reaction generating minor isomers (Table 2, entries 4, 6, and 12), the major adduct syn-4 was easily separated from its minor isomers by chromatography purification. The exception was the reaction with imines 6d and 6j, where the major adducts 4g and 4m were isolated, respectively, as an inseparable mixture contaminated with one minor isomer (Table 2, entries 3 and 9).

The *N*,O-protected isoseric acid and β -lactam were commonly used for the *semi-synthsis* of taxol.^{2b,4l,12} With adduct **4c** in hands, our attention was then focused on preparing the isoseric acid **9** and β -lactam **10** by selective manipulating of the protecting groups in **4c**. Acid **9** and β -lactam **10** were readily synthesized by the following two steps. Thus, hydrogenolysis of **4c** under 1 atm of hydrogen pressure for 12 h in the presence of 5% Pd(OH)₂ (Pearlman's catalyst) afforded **9** in 94% yield. When **9** was treated with 1.5 equiv of 2-chloro-1-methylpyridinium iodide and Et₃N in MeCN at 60 °C for 6 h, the β -lactam **10** was obtained in 45% yield as well as some unidentified byproducts (Scheme 3).





 a Reagents and conditions: (a) 10% Pd(OH)_2, 1 atm H_2 in MeOH, 12 h, 94% yield. (b) 1.5 equiv of 2-chloro-1-methylpyridinium iodide and Et_3N in MeCN at 60 °C, 45% yield.



FIGURE 2. Plausible mechanism for the lithium enolate imine addition.

To provide a rationale for the observed excellent diastereoselectivity using lithium enolate of O-Boc- α -hydroxyacetate 11 as a nucleophile, a tentative mechanism was suggested (Figure 2), which adopted the six/four-membered bicyclic transition state proposed by Davis, Ellman, and co-workers in their asymmetric addition reaction of lithium enolate to sulfinylimine.^{6a,d} Both trans-enolate 11 and imine 6 were linked together through lithium chelation to form six/four-membered bicyclic transition states TS-1 and TS-2. TS-1 should be the favored low-energy transition state, since the cis relationship between *a*-orientation R and *e*-orientation R^1 in **TS-1** has less steric hindrance and allows the easier access of C-2 to C-3, compared with the trans relationship between both e-orientation R and R¹ in **TS-2**. With the size increase of R¹ from methoxy carbonyl group to tertbutoxy carbonyl group, the addition reaction should be more likely to occur through TS-1 to afford syn-4 as the dominating adduct. The suggested mechanism explained well our experimental observation (Table 1, entries 1-3).

Conclusion

We have described that the enantiopure 3-substituted isoserine **4** with a (S_R , 2R, 3S) absolute stereochemistry can be readily prepared in high to excellent yields by the diastereoselective enolate addition of *O*-Boc- α -hydroxyacetate **5** with (S_R)-*tert*-butanesulfinylimine **6**. To the best of our knowledge, the current method provides the most straightforward access to enantiopure 3-substituted isoserine **4**, which is useful for the semi-syntheses of taxol and its derivatives.

Experimental Section

General Procedure for the Addition of O-Protected α -Hydroxyacetate 5 with (*S_R*)-*tert*-Butanesulfinylimine 6. Under N₂, to a solution of *O*-protected α -hydroxyacetate 5 (10.0 mmol) in 25 mL of dry THF was added a solution of base (10.0 mmol) at -78°C. After being stirred for 1 h at -78 °C, the mixture was added to (*S_R*)-*tert*-butanesulfinylimine 6 (2.0 mmol in 5 mL of dry THF and was stirred at -78 °C for 1-6 h until imine 6 was completely consumed and checked by TLC analysis. The reaction mixture was

⁽¹¹⁾ Cragg, G. M.; Newman, D. J. J. Nat. Prod. 2004, 67, 232.
(12) Denis, J. N.; Greene, A. E.; Guenard, D.; Gueritte-Voegelein, F.; Mangatal, L.; Potier, P. J. Am. Chem. Soc. 1988, 110, 5917.

quenched at -78 °C with saturated NH₄Cl solution and extracted with ethyl acetate (3 × 20 mL). The combined organic phases were dried with anhydrous MgSO₄. The solvent was removed under vacuum, and the residue was purified by chromatography to give adduct **4** and other isomers. The results were shown in Tables 1 and 2.

(*S_R*,*2R*,*3S*)-*N*-*tert*-Butanesulfinyl-*O*-methoxycarbonyl-3-phenylisoserine Benzyl Ester 4a: Purified by chromatography (acetone/petroleum 1:4, *R_f* 0.4) as viscous oil, ¹H NMR (400 MHz, CDCl₃) δ 7.28–7.41 (m, 10H), 5.27 (d, *J* = 2.8 Hz, 1 H), 5.21 (d, *J* = 12.4 Hz, 1H), 5.14 (d, *J* = 12.4 Hz, 1H), 5.03 (dd, *J* = 10.0, 2.8 Hz, 1H), 4.13 (d, *J* = 9.6 Hz, 1H), 3.76 (s, 3H), 1.14 (s, 9H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 167.1, 154.6, 137.7, 134.67, 134.60, 128.58, 128.52, 128.4, 128.3, 128.2, 127.0, 78.6, 67.6, 60.2, 56.6, 55.2, 22.3 ppm; HRMS-ESI calcd for C₂₂H₂₇N₁Na₁O₄S₁ (M + Na)⁺ 456.1457, found 456.1451.

Isomer of 4a: Purified by chromatography (acetone/petroleum 1:4, $R_f 0.35$) as viscous oil, ¹H NMR (400 MHz, CDCl₃) δ 7.19–7.39 (m, 10H), 5.39 (d, J = 5.2 Hz, 1H), 5.10 (d, J = 12.4 Hz, 1H), 5.03 (d, J = 12.4 Hz, 1H), 4.91 (t, J = 5.2 Hz, 1H), 4.22 (d, J = 4.8 Hz, 1H), 3.79 (s, 3H), 1.16 (s, 9H) ppm; ¹HRMS-ESI calcd for C₂₂H₂₇N₁Na₁O₄S₁ (M + Na)⁺ 456.1457, found 456.1454.

(*S_R*,2*R*,3*S*)-*N*-*tert*-Butanesulfinyl-*O*-Cbz-3-phenylisoserine Benzyl Ester 4b: Purified by chromatography (acetone/petroleum 1:7, *R_f* 0.2) as viscous oil, IR (KBr) 3300, 2959, 1750, 1498, 1455, 1385, 1270, 1242, 1073, 1040 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.24–7.40 (m, 15H), 5.32 (d, *J* = 2.8 Hz, 1H), 5.18 (d, *J* = 12.4 Hz, 1H), 5.15 (d, *J* = 12.4 Hz, 1H), 5.12 (d, *J* = 12.4 Hz, 1H), 5.09 (d, *J* = 12.4 Hz, 1H), 5.03 (dd, *J* = 9.6, 2.8 Hz, 1H), 4.12 (d, *J* = 9.6 Hz, 1H), 1.12 (s, 9H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 167.2, 154.1, 137.7, 134.7, 128.6, 128.58, 128.53, 128.4, 128.21, 128.13, 127.1, 78.7, 70.3, 67.6, 60.3, 56.7, 22.6, 22.5 ppm; HRMS-ESI calcd for C₂₈H₃₁N₁Na₁O₆S₁ (M + Na)⁺ 532.1770, found 532.1764.

(*S_R*,2*R*,3*S*)-*N*-*tert*-**Butanesulfinyl**-*O*-**Boc**-3-**phenylisoserine Benzyl Ester 4c:** Purified by chromatography (acetone/petroleum 1:8, *R_f* 0.2) as a white solid, recrystallized from acetone and petroleum; X-ray (see CIF file); mp 148–149 °C; $[\alpha]^{25}_{\rm D}$ +1.0° (*c* 1.0, EtOH); IR (KBr) 3296, 2975, 1746, 1497, 1458, 1426, 1395, 1366, 1291, 1243, 1208, 1160, 1113, 1046, 1009, 956, 862 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.22–7.41 (m, 10H), 5.27 (d, *J* = 0.8 Hz, 1H), 5.17 (s, 2H), 5.01 (d, *J* = 9.6 Hz, 1H), 4.14 (d, *J* = 9.6 Hz, 1H), 1.39 (s, 9H), 1.15 (s, 9 H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 167.5, 152.1, 137.8, 134.6, 128.5, 128.4, 128.0, 127.1, 83.3, 77.6, 67.4, 60.4, 56.6, 27.4, 22.3 ppm; Anal. Calcd for C₂₅H₃₃N₁O₆S₁: C, 63.13; H, 6.99; N, 2.95. Found: C, 63.08; H, 6.98; N, 3.08; HRMS-ESI calcd for C₂₅H₃₃N₁Na₁O₆S₁ (M + Na)⁺ 498.1926, found 498.1921.

(*S_R*,2*R*,3*S*)-*N*-*tert*-Butanesulfinyl-*O*-Boc-3-phenylisoserine Methyl Ester 4d: Purified by chromatography (acetone/petroleum 1:8, *R_f* 0.2) as a white solid, mp 85–87 °C; [α]²⁵_D +8.6° (*c* 1.2, CHCl₃); IR (KBr) 3345, 2957, 1745, 1499, 1461, 1393, 1370, 1352, 1310, 1280, 1162, 1136, 1080, 1018, 951 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.28–7.43 (m, 5H), 5.25 (d, *J* = 2.8 Hz, 1H), 5.01 (dd, *J* = 10.0, 2.8 Hz, 1H), 4.16 (d, *J* = 10.0 Hz, 1H), 3.78 (s, 3H), 1.41 (s, 9H), 1.19 (s, 9H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 168.1, 152.2, 138.0, 128.6, 128.1, 127.1, 83.4, 60.3, 56.7, 52.4, 27.5, 22.6, 22.4 ppm; Anal. Calcd for C₁₉H₂₉N₁O₆S₁: C, 57.14; H, 7.32; N, 3.51. Found: C, 57.10; H, 7.30; N, 3.64; HRMS-ESI calcd for C₁₉H₂₉N₁Na₁O₆S₁ (M + Na)⁺ 422.1613, found 422.1608.

Correlation of the Absolute Configurations by Chemical Transformation. Adduct 4d (0.202 g, 0.5 mmol) was dissolved in 10 mL of 6 M HCl in MeOH overnight. The mixture was poured into an ice-cooled saturated NaHCO₃ solution and was extracted with CH_2Cl_2 (3 × 10 mL). The organic phases were combined and dried with anhydrous Na₂SO₄. The solvent was removed under vacuum to afford 0.098 g of crude 7 (96% yield), which was pure enough for the next step. The crude 7 was dissolved in 15 mL of THF. To the THF solution, saturated NaHCO₃ (5 mL) and benzoyl

chloride (1.0 mmol) were added successively at 0 °C. After being stirred for 2 h at 0 °C, the reaction mixture was extracted with CH₂Cl₂ (3×10 mL). The combined organic phases were dried, and the solvent was removed under vacuum to give a residue, which was purified by chromatography to give **8** (0.123 g, 82% overall yield two steps).

(2*R*,3*S*)-3-Phenylisoserine Methyl Ester 7. The analytic sample was purified by chromatography (acetone/petroleum 1:4, R_f 0.2) as a white solid, mp 103–105 °C, [α]²⁵_D +10.6° (*c* 1.0, MeOH); IR (KBr) 3367, 3308, 3070, 2855, 2707, 1730, 1449, 1434, 1365, 1280, 1198, 1160, 1082, 922 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.41–7.26 (m, 5H), 4.30 (m, 2H), 3.80 (s, 3H), 2.00 (s, br, 2H)-ppm; ¹³C NMR (100 MHz, CDCl₃) δ 173.6, 141.9, 128.4, 127.5, 126.6, 75.0, 57.8, 52.3 ppm.

(2*R*,3*S*)-*N*-Benzoyl-3-phenylisoserine Methyl Ester 8: Purified by chromatography (acetone/petroleum 1:4, R_f 0.5) as a white solid, mp 183–184 °C, $[\alpha]^{25}_{\rm D}$ –48.8 (*c* 1.3, MeOH) {lit.^{1d} $[\alpha]^{23}_{\rm D}$ –49.6 (MeOH), lit.⁹ $[\alpha]^{23}_{\rm D}$ –48.0 (MeOH)}. ¹H NMR was identical to literature report.^{1d,9} IR (KBr) 3391, 3367, 1735, 1639, 1580, 1521, 1489, 1449, 1292, 1261, 1208, 1111 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.77 (m, 2H), 7.27–7.56 (m, 8H), 6.98 (d, *J* = 8.8 Hz, 1H), 5.74 (dd, *J* = 9.2, 0.8 Hz, 1H), 4.64 (dd, *J* = 4.0, 2.4 Hz, 1H), 3.85 (s, 3H), 3.27 (d, *J* = 3.6 Hz, 1H) ppm.

Synthesis of β **-Lactam 10.** Adduct **4c** (1.255 g, 2.6 mmol) in 20 mL of MeOH was subjected to hydrogenolysis under 1 atm hydrogen pressure in the presence of 10% of Pd(OH)₂ for 12 h. After we removed the Pd(OH)₂ by filtration, we evaporated the mixture to dryness. The residue was purified by chromatography (CH₂Cl₂/MeOH/AcOH 10:1:0.1, R_f 0.3) to afford acid **9** (0.931 g, 93% yield). Under N₂, a mixture of acid **7** (0.820 g, 2.1 mmol), Et₃N (0.323 g, 3.2 mmol), and 2-chloro-1-methylpyridinium iodide (0.800 g, 3.2 mmol) in 15 mL of dry MeCN was refluxed for 6 h. The solvent was removed by evaporation, and the residue was subjected to chromatography (acetone/petroleum 1:8, R_f 0.2) to give β -lactam **10** (0.347 g, 45% yield).

 $(S_R, 2R, 3S)$ -*N-tert*-Butanesulfinyl-*O*-Boc-3-phenylisoserine Acid 9: mp 144–146 °C; $[\alpha]^{25}_{D}$ +23.0° (*c* 1.2, CHCl₃); IR (KBr) 3357, 2980, 1741, 1610, 1452, 1372, 1351, 1309, 1279, 1249, 1229, 1166, 1132, 1098, 1006 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.40 (d, *J* = 7.6 Hz, 2H), 7.27 (t, *J* = 7.6 Hz, 2H), 7.22 (m,1H), 6.32 (br, 1H), 4.85 (s, 1H), 4.70 (dd, *J* = 9.2, 4.0 Hz, 1H), 1.34 (s, 9H), 1.09 (s, 9H)ppm; ¹³C NMR (100 MHz, DMSO-*d*₆) δ 170.0, 152.8, 140.9, 128.2, 127.9, 127.1, 80.9, 78.8, 67.2, 58.8, 56.2, 27.7, 25.3, 22.6 ppm; Anal. Calcd for C₁₈H₂₇N₁O₆S₁: C, 56.08; H, 7.06; N, 3.63. Found: C, 55.71; H, 7.06; N, 3.85; HRMS-ESI calcd for C₁₈H₂₇N₁Na₁O₆S₁ (M + Na)⁺ 408.1457, found 408.1451.

(*S_R*,*3R*,*4S*)-*N*-*tert*-Butanesulfinyl-3-(*tert*-butoxycarbonate)-4phenyl-β-lactam 10: White solid, mp 125–127 °C; $[\alpha]^{25}_{\rm D}$ +56.2° (*c* 1.2, CHCl₃); IR (KBr) 2979, 1792, 1745, 1458, 1369, 1278, 1252, 1130, 1101, 1047, 1014 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.33– 7.45 (m, 5H), 5.74 (d, *J* = 6.0 Hz, 1H), 5.36 (d, *J* = 6.0 Hz, 1H), 1.16 (s, 9H), 1.03 (s, 9H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 167.8, 150.6, 133.0, 129.7, 129.1, 128.4, 83.6, 57.7, 57.3, 27.1, 22.3 ppm; Anal. Calcd for C₁₈H₂₅NO₅S: C, 58.83; H, 6.86; N, 3.81. Found; C, 58.77; H, 6.94; N, 3.86; HRMS-ESI calcd for C₁₈H₂₅N₁-Na₁O₅S₁ (M + K)⁺ 406.1091, found 406.1098.

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Supporting Information Available: ¹H NMR and ¹³C NMR spectra of compounds 4, 7-10, and X-ray crystallographic data for 4c as a CIF file. This material is available free of charge via the Internet at http://pubs.acs.org.

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